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# Construction of Phenanthridinone Skeletons through Palladium-**Catalyzed Annulation**

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ABSTRACT: Herein, a straightforward synthetic approach for the construction of phenanthridin-6(5H)-one skeletons is disclosed. The developed protocol relies on palladium catalysis, providing controlled access to a range of functionalized phenanthridin-6(5H)-ones in 59-88% yields. Furthermore, plausible reaction pathways are proposed based on mechanistic experiments.

## INTRODUCTION

Phenanthridin-6(5H)-one represents a class of tricyclic Nheterocycles that is frequently encountered in alkaloids, such as phenaglydon, crinasiadine, and trisphaeridine (Figure 1, top). These compounds have been documented to possess biological and pharmaceutical activities, including antimycobacterial,<sup>1</sup>

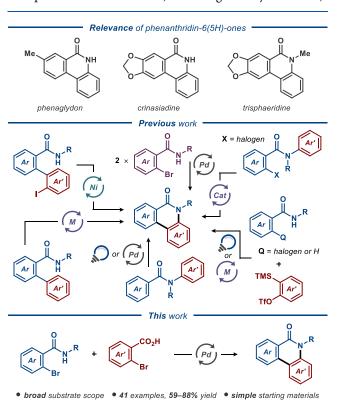


Figure 1. Relevance and synthetic approaches to phenanthridin-6(5H)-one derivatives.

antagonistic,<sup>2</sup> antiproliferative,<sup>3</sup> and antitubercular activities.<sup>4</sup> Significant attention has been devoted to developing novel synthetic methods for the construction of phenanthridin-6(5H)-one derivatives (Figure 1, middle). The Yamada group demonstrated the synthesis of phenanthridin-6(5H)-ones through nickel-catalyzed amidation of aryl iodides. At the same time, Chaudhary and co-workers disclosed an organocatalytic protocol proceeding through direct C(sp<sup>2</sup>)-H bond arylation. Similarly, phenanthridin-6(5H)-one derivatives have also been accessed in high yields using the free radical initiator AIBN<sup>9</sup> or microwave irradiation. Furthermore, phenanthridin-6(5H)-one derivatives have been efficiently assembled from 2-bromophenylbenzamides through a palladium-catalyzed process involving aryl-aryl coupling and deamidation.<sup>11</sup> Various strategies have utilized the oxidative coupling of benzamides to construct phenanthridin-6(5H)-one scaffolds. These annulation approaches do not require ortho-halogenation and have been realized with transition-metal-catalyzed 12 or photoinduced<sup>13</sup> manifolds. In recent years, a direct ortho-C-H/N-H annulation was developed to yield phenanthridin-6(5H)-one derivatives from benzamide and the aryne precursor 2-(trimethylsilyl)phenyl trifluoromethanesulfonate using O<sub>2</sub> or K<sub>2</sub>S<sub>2</sub>O<sub>8</sub> as oxidizing agents.<sup>14</sup>

It has been demonstrated that 2-bromobenzoic acid can be easily converted to the corresponding aryne in the presence of a Pd catalyst.<sup>15</sup> However, the generated aryne quickly undergoes a trimerization reaction to yield triphenylenes. In this context, we recently reported that 2-(2-bromophenyl)-1Hbenzo[d]-imidazole derivatives can be harnessed as an effective

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Table 1. Optimization of Reaction Conditions<sup>a</sup>

entry	[M]	ligand	base	solvent	temp (°C)	yield (%) <sup>b</sup>
1	CuI	_	$K_2CO_3$	DMF	100	0
2	AgOTf	_	$K_2CO_3$	DMF	100	0
3	$Pd(OAc)_2$	_	$K_2CO_3$	DMF	100	54
4	PdCl <sub>2</sub>	_	$K_2CO_3$	DMF	100	31
5	$Pd(PPh_3)_2Cl_2$	_	$K_2CO_3$	DMF	100	40
6	$Pd(PPh_3)_4$	_	$K_2CO_3$	DMF	100	48
7	$Pd(OAc)_2$	$PPh_3$	$K_2CO_3$	DMF	100	70
8	$Pd(OAc)_2$	Xantphos	$K_2CO_3$	DMF	100	68
9	$Pd(OAc)_2$	$P(4-MeOC_6H_4)_3$	$K_2CO_3$	DMF	100	64
10	$Pd(OAc)_2$	$P(4-MeC_6H_4)_3$	$K_2CO_3$	DMF	100	69
11	$Pd(OAc)_2$	$PPh_3$	$Na_2CO_3$	DMF	100	64
12	$Pd(OAc)_2$	$PPh_3$	$Cs_2CO_3$	DMF	100	72
13	$Pd(OAc)_2$	$PPh_3$	<sup>t</sup> BuOK	DMF	100	67
14 <sup>c</sup>	Pd(OAc) <sub>2</sub> /CuI	$PPh_3$	$Cs_2CO_3$	DMF	110	0
15	$Pd(OAc)_2$	$PPh_3$	$Cs_2CO_3$	DMF	80	57
16	$Pd(OAc)_2$	$PPh_3$	$Cs_2CO_3$	DMF	110	73
17	$Pd(OAc)_2$	PPh <sub>3</sub>	$Cs_2CO_3$	DMF	120	75
18	$Pd(OAc)_2$	$PPh_3$	$Cs_2CO_3$	DMF	130	73
19	$Pd(OAc)_2$	$PPh_3$	$Cs_2CO_3$	DMSO	120	73
20	$Pd(OAc)_2$	$PPh_3$	$Cs_2CO_3$	DMA	120	72
21	$Pd(OAc)_2$	$PPh_3$	$Cs_2CO_3$	xylene	120	65
22	$Pd(OAc)_2$	$PPh_3$	$Cs_2CO_3$	toluene	120	70
23	-	$PPh_3$	$Cs_2CO_3$	DMF	120	0

"Reaction conditions: Reactions were carried out with 1a (107 mg, 0.50 mmol), 2a (121 mg, 0.60 mmol), catalyst (10 mol %), ligand (20 mol %), and base (1.0 mmol) in solvent (5.0 mL) under argon for 10 h. "Isolated yields of 3a after purification by column chromatography. "Reaction was carried out with 1a (107 mg, 0.50 mmol), 2a (121 mg, 0.60 mmol), Pd(OAc)<sub>2</sub> (5 mol %), CuI (10 mol %), PPh<sub>3</sub> (20 mol %), and Cs<sub>2</sub>CO<sub>3</sub> (0.5 mmol) in DMF (5.0 mL) under argon for 8 h.

coupling partner in combination with 2-bromobenzoic acids to give the corresponding N-fused (benzo)imidazophenanthridine scaffolds in high yields. In continuation of our previous studies directed to transition-metal-assisted synthesis of heterocycles,  $^{17}$  we envisaged that phenanthridin-6(5H)-one derivatives could be directly assembled from N-substituted 2-bromobenzamides 1 and 2-bromobenzoic acids 2 in the presence of a metal catalyst (Figure 1, bottom).

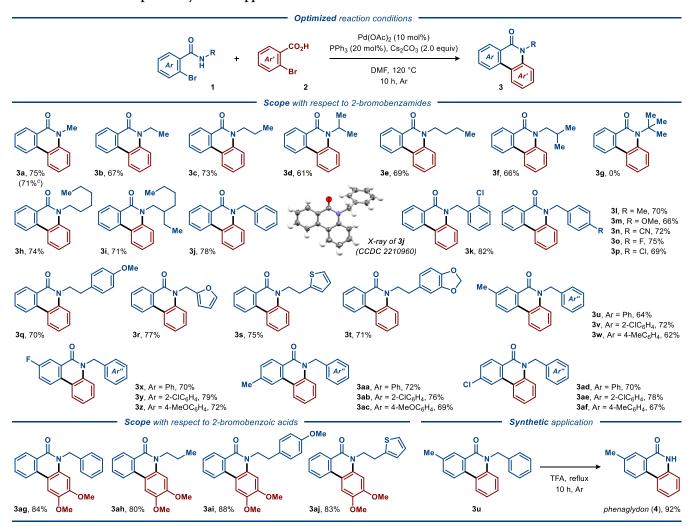
## RESULTS AND DISCUSSION

We commenced our investigation by utilizing 2-bromo-Nmethylbenzamide (1a) and 2-bromobenzoic acid (2a) as the model substrates, CuI as the catalyst precursor, and K<sub>2</sub>CO<sub>3</sub> as the base in DMF at 100 °C. To our disappointment, the desired product 3a was not detected under these reaction conditions (Table 1, entry 1). A similar outcome was observed with AgOTf as the metal catalyst (Table 1, entry 2). Gratifyingly, formation of the desired annulation product 3a could be promoted by palladium-based catalysts, including Pd(OAc)<sub>2</sub>, PdCl<sub>2</sub>, Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub>, and Pd(PPh<sub>3</sub>)<sub>4</sub> (Table 1, entries 3-6), with Pd(OAc)<sub>2</sub> displaying the best reactivity and furnishing the desired product in 54% yield (Table 1, entry 3). Notably, the addition of auxiliary phosphine-based ligands, such as PPh<sub>3</sub>, Xantphos, P(4-MeOC<sub>6</sub>H<sub>4</sub>)<sub>3</sub>, and P(4-MeC<sub>6</sub>H<sub>4</sub>)<sub>3</sub>, promoted the desired reactivity (Table 1, entries 7-10) with PPh<sub>3</sub> providing product 3a in 70% yield (Table 1, entry 7). Apart from  $K_2CO_3$ , other common bases, such as  $Na_2CO_3$ ,

Cs<sub>2</sub>CO<sub>3</sub>, and <sup>t</sup>BuOK, were evaluated and found less critical for the desired transformation (Table 1, entries 11-13). Carrying out the reaction under the optimized conditions for our previously disclosed protocol<sup>16</sup> for the synthesis of N-fused (benzo)imidazophenanthridine scaffolds did not afford the desired annulation product 3a (Table 1, entry 14). Instead, the trimerization product (triphenylene) was afforded under these reaction conditions. Next, the effect of the reaction temperature was examined (Table 1, entries 15-18) with 120 °C being the most suitable for the developed protocol. The use of polar aprotic solvents, such as DMF, DMSO, and DMA, was revealed to be beneficial (Table 1, entries 17, 19-20), while the nonpolar solvents xylene and toluene resulted in slightly diminished yields (Table 1, entries 21-22). Finally, a control experiment conducted in the absence of Pd(OAc)<sub>2</sub> highlights the critical role of the palladium precursor in achieving effective coupling (Table 1, entry 23).

After the optimal reaction conditions were identified, the scope and limitations of the developed protocol were evaluated. Initially, a series of *N*-substituted 2-bromobenzamides 1 were engaged in a reaction with 2-bromobenzoic acid 2a. Aliphatic groups, such as methyl, ethyl, "propyl, and "butyl, all furnished the corresponding products 3b—3f and 3h—3i in moderate to high yields (61—75%). However, *N*-'butyl-2-bromobenzamide failed to produce the desired annulation product 3g, presumably due to ample steric hindrance. The use of 2-bromobenzamides 1 bearing various *N*-substituted

Scheme 1. Reaction Scope and Synthetic Application a,b



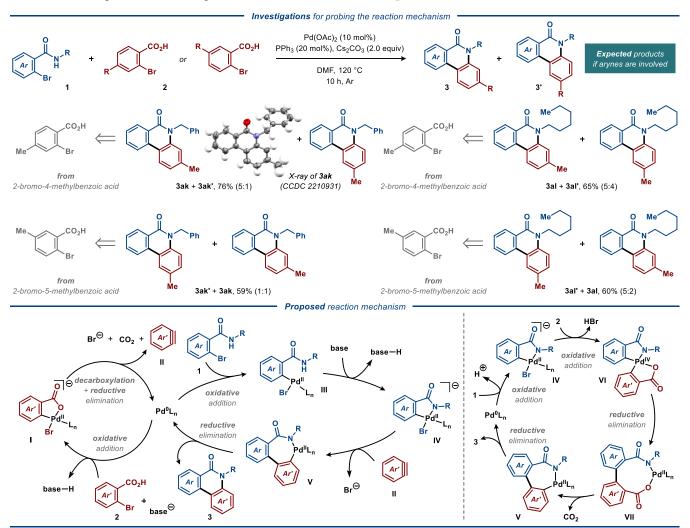
"Reaction conditions: Reactions were carried out with 1 (0.50 mmol), 2 (0.60 mmol), Pd(OAc)<sub>2</sub> (12 mg, 0.05 mmol), PPh<sub>3</sub> (26 mg, 0.10 mmol), and Cs<sub>2</sub>CO<sub>3</sub> (326 mg, 1.0 mmol) in DMF (5.0 mL) under argon at 120 °C for 10 h. <sup>b</sup>Isolated product yields are reported. <sup>c</sup>Reaction carried out on a 1 mmol scale.

aromatic and heteroaromatic moieties demonstrated that various functional groups, such as halogens, ethers, nitrile, furan, and thiophene, were compatible with the developed protocol, furnishing products 3j-3t in moderate to high yields (66-82%). The structure of product 3j was confirmed by single-crystal X-ray analysis (CCDC 2210960).

The synthetic versatility of the developed protocol was further explored by employing 2-bromobenzamides 1 with a range of substituents at the aromatic core (Scheme 1). The reactions with 2-bromobenzamides 1 substituted with various aliphatic, chloro, and fluoro groups all provided the expected annulation products 3u-3af in moderate to high yields (62-79%). Next, the scope of compatible 2-bromobenzoic acid annulation partners 2 was evaluated (Schemes 1 and 2). Here, 4,5-dimethoxy-2-bromobenzoic acid (2b) underwent effective annulation with N-substituted 2-bromobenzamides to produce 3ag-3aj in high yields (84-88%, Scheme 1). Finally, the disclosed protocol was successfully applied to access quinolone-derived alkaloid phenaglydon (4). Thus, subjecting annulation product 3u to refluxing trifluoroacetic acid afforded the debenzylated product phenaglydone (4) in an excellent yield of 92% (Scheme 1).

To probe the reaction mechanism, a set of control reactions were carried out under the optimized reaction conditions. When 4- or 5-substituted 2-bromobenzoic acids were used as the coupling partners, the respective annulated products were obtained as mixtures of two regioisomers (Scheme 2, top). Such poor regioselectivity indicates that the reaction proceeds through arynes as the key intermediates, as has been proposed for related transformations featuring palladium catalysis. <sup>18</sup> Based on the literature precedents, <sup>19</sup> a plausible mechanism that does not contradict the above control reactions is proposed (Scheme 2, bottom left). Initially, base-assisted oxidative addition of 2-bromobenzoic acid 2 to Pd<sup>0</sup> provides the key aryl-Pd<sup>II</sup> species I. This species undergoes extrusion of CO<sub>2</sub> to afford aryne intermediate II while regenerating Pd<sup>0</sup> and completing the first of the catalytic cycles. Meanwhile, the second of the catalytic cycles is onset by oxidative addition of the Pd<sup>0</sup> catalyst to 2-bromobenzamide 1 to give aryl-Pd<sup>II</sup> species III, which in the presence of a base furnishes the five-membered palladacycle IV. Insertion of previously produced aryne II into the PdII-C bond of IV results in C-C bond formation, while subsequent reductive elimination from the seven-membered palladacycle V forges the desired

Scheme 2. Investigations for Probing the Reaction Mechanism and Proposed Reaction Mechanism<sup>a,b</sup>



"Reaction conditions: Reactions were carried out with 1 (0.50 mmol), 2 (0.60 mmol), Pd(OAc)<sub>2</sub> (12 mg, 0.05 mmol), PPh<sub>3</sub> (26 mg, 0.10 mmol), and Cs<sub>2</sub>CO<sub>3</sub> (326 mg, 1.0 mmol) in DMF (5.0 mL) under argon at 120 °C for 10 h. <sup>b</sup>Isolated product yields are reported. Regioisomeric ratios were determined by <sup>1</sup>H NMR analysis.

C-N bond. Thereby, the latter step regenerates the Pd<sup>0</sup> catalyst, concluding the second of the catalytic cycles, and furnishes the desired annulation product 3. An alternative mechanism proceeding without formation of an aryne intermediate features a single catalytic cycle and Pd<sup>IV</sup> species as the key intermediate (Scheme 2, bottom right). 20 Here, the reaction is onset by oxidative addition of 2-bromobenzamide 1 to the Pd<sup>0</sup> catalyst, furnishing aryl-Pd<sup>II</sup> intermediate IV. In the key step of the reaction, this intermediate undergoes a second oxidative addition reaction to 2-bromobenzoic acid 2, producing diaryl-Pd<sup>IV</sup> species VI. Subsequently, this species undergoes reductive elimination to produce the biaryl PdIImetallacycle VII, which eliminates CO<sub>2</sub> to furnish the Pd<sup>II</sup> intermediate V. Finally, the latter intermediate undergoes reductive elimination, concluding the catalytic cycle and furnishing desired product 3.

## CONCLUSIONS

In conclusion, we disclosed a simple procedure for accessing phenanthridin-6(5H)-one derivatives through palladium-mediated annulation of 2-bromobenzamides and 2-bromobenzoic

acids. The annulation reaction delivers the phenanthridin-6(5H)-one derivatives in high yields and is compatible with a variety of functional groups, providing a modular method for accessing a range of structurally diversified phenanthridin-6(5H)-one motifs.

# EXPERIMENTAL SECTION

**General Information.** All reagents were purchased from commercial sources and used without treatment, unless otherwise indicated. The products were purified by column chromatography over silica gel. <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were recorded at 25 °C on a Varian spectrometer at 400 and 101 MHz, respectively, with TMS as the internal standard. High-resolution mass spectra (HRMS) were recorded on a BRUKER AutoflexIII Smartbeam mass spectrometer. High-resolution mass spectra (HRMS) were recorded on a Bruker microTof using electrospray ionization (ESI).

**General Procedure for the Synthesis of Phenanthridinones 3.** To a 10 mL Schlenk tube equipped with a magnetic stir bar were added 2-bromobenzamide **1** (0.500 mmol, 1.00 equiv), obromobenzoic acid **2** (0.750 mmol, 1.50 equiv), DMF (4.0 mL), Cs<sub>2</sub>CO<sub>3</sub> (163 mg, 0.500 mmol, 1.00 equiv), PPh<sub>3</sub> (26 mg, 0.100 mmol, 0.200 equiv), and Pd(OAc)<sub>2</sub> (11 mg, 0.05 mmol, 0.100 equiv). The reaction mixture was stirred at 120 °C in an oil bath for about 10

h. The resulting mixture was concentrated, and the residue was taken up in ethyl acetate. The organic layer was washed with brine, dried over  $Na_2SO_4$ , and concentrated. Purification of the crude product by column chromatography (silica gel; petroleum ether/ethyl acetate 30:1) afforded 3.

## ASSOCIATED CONTENT

## **Data Availability Statement**

The data underlying this study are available in the published article and its online Supporting Information.

# Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.joc.3c01429.

Experimental procedures, characterization data, and copies of NMR spectra for all obtained products (PDF)

FAIR data, including the primary NMR FID files, for compounds 3a-3z, 3aa-3al, and 4 (ZIP)

#### **Accession Codes**

CCDC 2210931 and 2210960 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via <a href="www.ccdc.cam.ac.uk/data\_request/cif">www.ccdc.cam.ac.uk/data\_request/cif</a>, or by emailing <a href="data\_request@ccdc.cam.ac.uk">data\_request/cif</a>, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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#### Notes

The authors declare no competing financial interest.

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